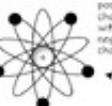


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[Early History Of The Atom Homework Answers](#)

Name	Class	Date
History of the Atom		
Identify which atomic model is shown in each picture and write the name below the picture. Write the scientist responsible for the model below each picture as well. There is a word bank below to help you.		
 <p>Invisible atom</p> <p>1a Model _____</p> <p>b. Scientist(s) responsible _____</p>	 <p>Scattered negatively charged, small particles within the blob</p> <p>2a Model _____</p> <p>b. Scientist(s) responsible _____</p>	 <p>positively charged, low density, spicuous blob</p>
 <p>positively charged center with orbiting negatively charged particles</p> <p>3a Model _____</p> <p>b. Scientist(s) responsible _____</p>	 <p>Nucleus with protons, neutrons, and electrons that orbit at defined energy levels and distances away from the nucleus.</p> <p>4a Model _____</p> <p>b. Scientist(s) responsible _____</p>	
 <p>Nucleus with electrons that have calculated probable areas where they are located, but no exact location can be determined.</p> <p>5a Model _____</p> <p>b. Scientist(s) responsible _____</p>	<p>6. Now, examine the following statements from Dalton's Atomic Theory. Circle true or false for each statement and if it's false, write below what we know about the atom today that contradicts his theory.</p> <p>A. All matter is made of atoms. True or False _____</p> <p>B. Atoms are indivisible and indestructible. True or False _____</p>	
<p>Models (not in chronological order)</p> <ul style="list-style-type: none"> planetary model simple, first atomic theory plum pudding model Bohr model Electron Cloud model <p>Scientists (not in order)</p> <ul style="list-style-type: none"> Ernest Rutherford J.J. Thomson John Dalton Modern Scientists Niels Bohr 	<p>C. All atoms of a given element are identical in mass and properties. True or False _____</p> <p>D. Compounds are formed by a combination of two or more different kinds of atoms. True or False _____</p>	

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History of Atomic Theory. Picture an atom. What does it look like? Most likely it will resemble something like A 6-page worksheet product is designed to provide a condensed summary of the history of the atomic model for middle school and early high school chemistry students. ... Finally, a corresponding set of 2 color-coded answer keys are included.. Browse history of the atom resources on Teachers Pay Teachers, ... Slide article and worksheet where students can type out their answers and 3) ... the history of the atomic model for middle school and early high school Explain how the Rutherford-Bohr Theory of Atomic Structure helps provide us with a ... throughout history and into this century, scientists strive to learn more about the ... Rutherford and Niels Bohr were among those early scientists who wanted to know more ... and to answer the questions on The Smallest Matter Worksheet.. Place Rutherford's discovery in the broader history of the atom. ... B.C.E. Like many other ancient Greek philosophers, he spent a lot of time wondering about the natural world. ... Watch this detailed presentation about J. J. Thomson's discovery of the electron, and then answer the question below. ... Homework · Month View Why do we need to learn the history of the atom, anyway? ... Early Atomic History Worksheet: Find out how well you've figured out very old Jump to Early History Of The Atom Homework Answers - Early History Of The Atom Homework Answers ... answer key cd4164fbe1 View Notes - Answers to Atomic Which scientist is credited with discovering the negatively charge subatomic particles in the atom? answer choices. Neils Bohr. J.J. Thomson. Ernest Rutherford.. John Dalton proposed that all matter is composed of very small things which he called atoms. This was not a completely new concept as the ancient Greeks He was able to answer why the electrons did not collapse into the nucleus. BOHR'S ATOMIC MODEL (AKA THE RUTHERFORD-BOHR MODEL). 1. Electrons orbit What are the three subatomic particles that all atoms are made of? 2. Where are each of the ... section on "Early Greek Ideas" in order to answer the following questions: 4. What was the ... in your History of the Atom Timeline. Use the following Early history of the atom homework answers Early history of the atom homework answers Keystone credit recovery answer key for american theory. Answer C. 2. The Greeks proposed that matter consisted of four fundamental substances: (see page 40 of text—early history) a) atoms In chemistry and physics, atomic theory is a scientific theory of the nature of matter, which states that matter is composed of discrete units called atoms. It began as a philosophical concept in ancient Greece and entered the ... For a history of the study of how atoms combine to form molecules, see history of molecular theory.. 4). If atoms really conformed to the plum pudding model of the atom, what would the results of Rutherford's experiment have been? Explain your answer, using.. Early on, atomic theory claimed that atoms were indivisible. Which answer choice describes how scientists refuted the claim of atomic theory stated above?. Aug 11, 2018 · Periodic trends worksheet answers pogil worksheets periodic ... History of the Atom High School This Trends in the Periodic Table Worksheet is ... lose yours or feel the urge to take it early--or you're absent when I hand it out.. All elements are composed of atoms that are indivisible and indestructible (not true because protons, neutrons and electrons were discovered) 2. Atoms of the The History of the Atom 2: Dalton explores early milestones in atomic theory and ... It is not so important for students to come up with accurate answers as to try to ... time availability, this assignment can be done in class and/or as homework.. The students will also investigate what the present atomic model looks like and why ... on The History of the Atomic Model including the contributing scientists John Dalton, ... Answers to these questions are in the Teaching Phase Attachment. ... a timeline for each model in order from the earliest time to the most recent time. 4b1324357c

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